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STUDIES ON SIALIC ACIDS I. DETERMINATION OF ANOMERIC CONFIGURATION OF NEURAMINIC ACID DERIVATIVES BY CIRCULAR DICHROISM¹⁾ Haruo Ogura^{*} and Kimio Furuhata School of Pharmaceutical Sciences, Kitasato University Shirokane, Minato-ku, Tokyo 108, Japan

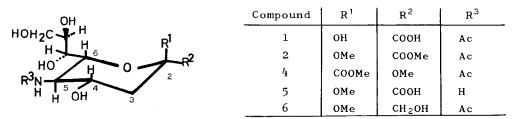
Summary: CD spectra were recorded for methyl α - and β -glycosides of D-neuraminic acid, and the band at the wave-length lower than 200 nm was attributed to the acetamido group. The Cotton effect at higher wave-length around 220 nm arose from the n- π^{*} transition of the carboxyl group. Thus α -linked glycosides showed a negative band, while β -glycosides gave arise to a positive band.

N-Acetyl-D-neuraminic acid (5-acetamido-3,5-dideoxy-D-glycero- β -D-galacto-2-nonulopyranosidic acid; 1) is widely distributed in the implication of membrane glycoproteins in animal cells. From the determination of the crystalline structure of the compound by X-ray analysis² and NMR studies,^{3,4} it has been shown that the conformation of the pyranose ring of 1 in the natural products is 1C and the configuration at C-2 is α .⁵

In this paper we wish to report the synthesis of methyl α - and β -glycosides of N-acetylneuraminic acid and some related derivatives, and to confirm their stereochemistry by means of NMR and CD studies.

N-Acetyl-D-neuraminic acid (1) was obtained from edible birds' nest substance in high yield (6.6%) by the method of Czarniecki and Thoruton.⁴⁾ Methyl β -glycoside of methyl 5-N-acetyl-D-neuraminate (2),⁶⁾ methyl α -glycoside of methyl 5-N-acetyl-D-neuraminate (4),^{7,8)} methyl β -glycoside of D-neuraminic acid (5),⁹⁾ and methyl 5-acetamido-3,5-dideoxy-D-glycero- β -D-galacto-2-nonulopyranoside (6)¹⁰⁾ were prepared following the procedures shown in the above literatures.

Table I. Derivatives of N-Acetyl-D-neuraminic Acid



Methyl 5-acetamido-3,5-dideoxy-<u>D</u>-<u>glycero</u>- α -<u>D</u>-<u>galacto</u>-2-nonulopyranoside (χ) was prepared from methyl α -glycoside of methyl 5-<u>N</u>-acetyl-4,7,8,9-tetra-<u>O</u>acetyl-<u>D</u>-neuraminate (3)⁸⁾ with sodium methoxide to yield $\frac{4}{4}$. This compound ($\frac{4}{4}$) was reduced with NaBH₄ in methanol and the reaction mixture was treated with Dowex-50 (H⁺ type) at -20° to yield 40% of 7 as colorless needles, mp 162-165°.¹¹⁾

Methyl α -glycoside of D-neuraminic acid (8) was prepared by treatment of $\frac{3}{2}$ with PCl₅, subsequently, $\overline{CH_3OH}$, and $Ba(OH)_2$.¹²⁾ The reaction mixture was treated with Dowex-2 (OH⁻ type) to yield 12% of 8 as a white powder.¹¹⁾

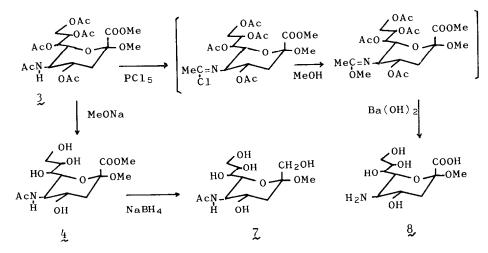


Figure 1 shows the CD spectra of two typical β - and α -methyl glycosides (2, 4). The large positive band at low wave-length at 194 nm was attributed to the acetamido group, while the smaller spectral features around 210-225 nm were assigned to the carboxyl chromophore.

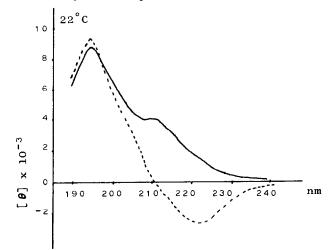


Figure 1. CD curves of $2 \pmod{4}$ and $4 \pmod{4}$

This was confirmed from CD data of the corresponding carboxy-reduced compound $(\underline{6}, \underline{7})$ summarized in Table II, in which the band at higher wavelength was lost. On the other hand, methyl β - and α -glycosides of \underline{D} -neuraminic acid $(\underline{5}, \underline{8})$ showed that the band at lower wave-length was lost.

Compound (Anomeric configuration)		Higher band (nm)	Molecular ellipticity [0]
(miomerre conriguration)	()		
5 (β)	-	215	+2040
6 (β)	193	-	+12600
7 (α)	193	-	+12730
8 (α)		216	-1520

Table II. CD Data of the Typical α_- and $\beta_-Glycosides~(H_2O,~22°C)$

In conclusion, the strong band at lower wave-length is attributed to the acetyl group and the sign of the Cotton effect is positive in both α - and β -anomers. In contrast, the sign of the n- $\pi^{\mathbf{x}}$ Cotton effect at higher wave-length is positive for β -glycosides $(\frac{1}{2}, \frac{2}{2}, \frac{5}{2})$ and negative for α -glycosides $(\frac{3}{2}, \frac{4}{2}, \frac{8}{2})$.

Figure 2 shows spectral changes with time which were observed for methyl β - and α -glycosides (2, 4). β -Form compound (2) in 1N HCl (methanol) shows a positive n- π^{*} Cotton effect at 222 nm ([θ] +2432) owing to the carboxyl group, and the figure of CD curve did not change after 20 hr. In contrast, α -form compound (4) in 1N HCl (methanol) shows a negative n- π^{*} Cotton effect at 213 nm ([θ] -4174) owing to the carboxyl group, and the figure of CD curve changed

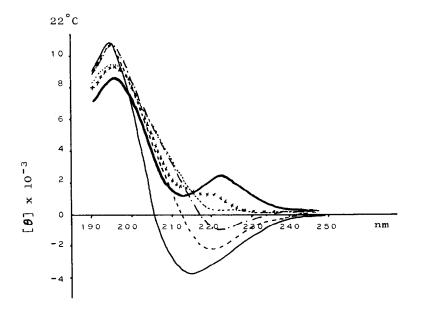


Figure 2. CD curves of $\frac{2}{2}$ (----) and $\frac{4}{4}$ (----; $\frac{4}{4}$ hr ---; $\frac{8}{4}$ hr; 20 hr ***)

by the elapse of time as shown in Figure 2. After 20 hr, the carboxyl negative band changed to positive and the curve approached that of β -glycoside (2) (70% of β -glycoside and 20% of α -glycoside were observed by the NMR spectrum)¹³⁾.

This evidence probably supports that the bonded N-acetyl-D-neuraminic acid in animal cells is α -glycoside, while the extracted N-acetyl-D-neuraminic acid is β -glycoside by the configurational change.

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REFERENCES

- 1) A part of this work was presented at the 23rd Symposium of the Chemistry of Natural Products, p. 295 (1980).
- A. M. O'Connel, <u>Acta Cryst.</u>, <u>B29</u>, 2320 (1973); J. L. Flippen, <u>Acta Cryst.</u>, <u>B29</u>, 1881 (1973); A. Biedl, <u>Naturwissenshaften</u>, <u>58</u>, 95 (1971).
- 3) P. Lutz, W. Lochinger, and G. Taigel, <u>Chem. Ber.</u>, <u>101</u>, 1089 (1968);
 U. Dabrowski, H. Friebolin, R. Brossmer, and M. Supp, <u>Tetrahedron Lett</u>., 4637 (1979).
- 4) M. F. Czarniecki and E. R. Thornton, <u>J. Am. Chem. Soc</u>., <u>99</u>, 8273 (1977).
- 5) A. Rosenberg and C. -L. Schengrund, Ed., "Biological Roles of Sialic Acid", Plenum Press, New York, N. Y., 1976; R. Schauer, <u>Angew. Chem. Internat.</u>, 12, 127 (1973).
- 6) R. K. Yu and R. Ledeen, <u>J. Biol. Chem</u>., <u>244</u>, 1306 (1969).
- 7) P. Meindl and H. Tuppy, Monats. Chem., 96, 802 (1965).
- 8) R. Kuhn, P. Lutz, and D. L. MacDonald, <u>Chem. Ber</u>., 99, 611 (1966).
- 9) R. Scheuer and H. -P. Buscher, <u>Biochim. Biophys. Acta</u>, 338, 369 (1974).
- 10) R. Brossmer and L. Holmouist, <u>Z. Physiol. Chem.</u>, <u>352</u>, 1715 (1971).
- All new compounds show satisfactory elemental analyses and spectral properties.
 - ¹H-NMR [60 MHz, D_2O , δ ppm (DSS)]
 - 7: 1.73 (1H, dd, <u>J</u>=11.0 and 13.0 Hz, $3-H_{ax}$), 2.21 (3H, s, NHCOC<u>H</u>₃), 2.29 (1H, dd, <u>J</u>=4.5 and 13.0 Hz, $3-H_{eq}$), 3.32 (3H, s, OCH₃).
 - 8: 1.70 (1H, dd, $\underline{J}=4.6$ and 13.0 Hz, $3-\underline{Hax}$), 2.55 (1H, dd, $\underline{J}=11.5$ and 13.0 Hz, $3-\underline{Heq}$), 3.40 (3H, s, OCH₃).
- 12) A. Koda, K. Takanobu, I. Isaka, T. Kashiwagi, K. Takahashi, S. Kawahara, and M. Murakami, <u>YAKUGAKU ZASSHI</u>, 92, 459 (1972).
- Isosbestic point is not observed in Figure 2, probably owing to form about 10% of unknown compound from the NMR spectrum.

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